Copper-Assisted/Copper-Free Synthesis of Functionalized Dibenzo $[b,f]$ oxepins and Their Analogs via a One-Pot Tandem Reaction

by Yuqin Wang, Yanhong Chen, Qian He, Yuyuan Xie, and Chunhao Yang*

State Key Laboratory of Drug Research, Shanghai Institute of Materia Medica, SIBS, Chinese Academy of Sciences, 555 Zu Chong Zhi Road, Shanghai 201203, P. R. China (e-mail: chyang@simm.ac.cn)

A simple, convenient, and efficient method for the formation of functionalized dibenzo $[b, f]$ oxepins and their analogs bearing both electron-donating and electron-withdrawing groups has been developed via a one-pot cascade reaction. Most starting materials are commercially available 2-(2-hydroxyphenyl) acetonitriles and 2-haloarylaldehydes. The procedure makes use of $Cs₂CO₃$ as the base, and DMF as solvent under copper-assisted/copper-free conditions. The reaction has a comprehensive group tolerance for substrates. Most of the reactions were complete within 1 h in good-to-excellent yields, and the reaction temperatures were relatively low. The protocol could be scaled up to grams without lowering the yield. A reaction mechanism was also proposed.

Introduction. – Dibenzo $[b, f]$ oxepin is an important framework in medicinal chemistry, and its derivatives occur in several medicinally relevant plants $[1-6]$. The scaffold has attracted considerable attention due to its diverse biological features such as anti-estrogenic [7], antidepressant $[7-9]$, analgesic [10], anti-inflammatory [3] [11] [12], antipsychotic [12–15], angiotensin II receptor antagonistic [16], antioxidant [17], antimycobacterial [4], antidiabetic [18], and antitumor activities [6], as well as anti-apoptosis [19] properties (Fig.). The treatment of progressive neurodegenerative diseases [20] such as *Parkinson*'s and *Alzheimer*'s diseases [21] with synthetic dibenzo[*b*,*f*]oxepine derivatives is of particular interest.

Considerable efforts have been made to synthesize the dibenzo $[b, f]$ oxepin skeleton since the report of Manske and Ledingham [22] [23]. The most widely applied approaches to the synthesis of the scaffold involved constructing substrates in multiple steps via the Ullmann coupling reaction and the Friedel–Crafts reaction (Scheme 1, *Examples A* [17] and *B* [10] [11] [20] [24]), but the total yields were not satisfying. *Yang* et al. reported a method via benzoin condensation to prepare dibenzo $[b, f]$ oxepine by using the hypertoxic reagent KCN, and obtained the product in low yield $(22-25%)$ [25]. Another reported strategy for the formation of this fused seven-membered ring involved the *Wagner–Meerwein* rearrangement of a number of structurally diverse 9- $(1-hydroxvalkyl)x$ anthenes (*Scheme 1, Example C* [26]). This procedure required 3-5 wt-equiv. of P_2O_5 , toxic benzotrifluoride as the solvent, and a long reaction time (24 – 96 h) at reflux temperature and gave the products generally in low-to-moderate yields $(20-56%)$. Recently, *Cong et al.* reported a Mn^{III}-based oxidative 1,2-radical rearrangement to form dibenzo[b,f]oxepins by using a large excess of $Mn(OAc)$ ₃ (4 molequiv.) in boiling glacial AcOH in moderate yields as well [27]. In 2001, Chernysheva *et al.* [28] reported a one-pot procedure to prepare NO₂-substituted dibenz[b,f]ox-

 $©$ 2013 Verlag Helvetica Chimica Acta AG, Zürich

Figure. Pharmacologically active dibenzo[b,f]oxepins

epines. However, this method involved explosive 2,4,6-trinitrotoluene as the starting reagent. These existing methods have some inherent disadvantages, i.e., i) multiple synthetic steps, ii) low total yields, iii) harsh and not environment-friendly conditions, and iv) long reaction times, and they cannot provide fast and efficient access to a library of functionalized dibenzo $[b, f]$ oxepins and thus limit the molecular diversity for lead optimization and drug screening. Considering the promising biological activities of $diberzo[*b*,*f*]oxepins, a facile, practical, and versatile method that enables quick access$ to dibenzo $[b, f]$ oxepins is highly desirable.

Here, we report a convenient method for preparing dibenzo $[b, f]$ oxepins with various functional groups via a one-pot cascade reaction in good-to-excellent yields under Cu-assisted as well as Cu-free conditions.

Results and Discussion. – We started with commercially available 2-bromobenzaldehyde $(2a)$ and 2- $(2$ -hydroxyphenyl)acetonitrile $(1a)$, which was prepared from 2- $(2-)$ methoxyphenyl)acetonitrile by deprotection of the methyl ether group with BBr_3 [29], as the model reaction. First, the reaction was catalyzed by CuI in dioxane and K_2CO_3 as the base at 100 $^{\circ}$ (*Table 1, Entry 1*). The desired compound 3a was obtained only in 20% yield. When DMF was used as the solvent, the reaction proceeded smoothly, resulting in an 88% yield (*Table 1, Entry 2*). Next, the bases were investigated. As shown in Table 1, the reaction went well with Cs_2CO_3 , and the yield of 3a was quite high (*Entry 6*). The yield was lower when DBU (=1.8-diazabicyclo[5.4.0]undec-7-ene) was used as base (Table 1, Entry 5), and the reaction was not complete at 100 $^{\circ}$ after 5 h. With a prolonged time (12 h) and higher temperature (150 $^{\circ}$), the reaction afforded 3a in 61% yield. The reaction did not work when EtONa and DMAP $(=4-(\text{dimethyl-}$ amino)pyridine) were used as bases (*Table 1, Entries 3* and 4). Then, the loading of CuI was investigated. When the reaction was conducted in the presence of an increased Scheme 1. Reported Synthesis of Dibenzo/b,f/oxepins

amount of CuI, the yield did not improve (*Table 1, Entry 7*). When the amount of CuI was decreased to 0.05 equiv. (*Table 1*, *Entry 8*), a comparable yield was obtained. To our delight, 0.01 equiv. of CuI (Table 1, Entry 9) was as effective without a prolonged reaction time. The reaction was also studied under a Cu-free condition (Table 1, Entry 14). Although 3a was obtained in 60% yield, the starting material was not completely consumed, and the yield decreased with prolonged time. Next, the amount of base was explored. By decreasing the amount of Cs_2CO_3 (Table 1, Entries 10 and 11), a slightly lower yield of 3a was obtained. Finally, when the reaction temperature was raised to 120 $^{\circ}$, the reaction was complete within 1 h, but the yield decreased (*Table 1*, *Entry 12*). The yield of 3a was only 66% when the reaction was conducted at 80 $^{\circ}$ (Table 1, Entry 13).

Under the optimal reaction conditions $(0.01 \text{ equiv. of CuI}, 3 \text{ equiv. of Cs}₂, CO₃)$ DMF, 100° , 1 h), we investigated the scope and limitations of the reaction employing a variety of ortho-haloarene-carbaldehydes and 2-(2-hydroxyphenyl)acetonitrile (1a). As compiled in Table 2, a number of 2-bromobenzaldehydes bearing both electrondonating (EDG) and electron-withdrawing groups (EWG) reacted very well. With 2 bromo-3-fluorobenzaldehyde (2b), the reaction proceeded smoothly with an excellent yield (*Table 2, Entry 1*). The effect of substitutions at $C(4)$ of 2-bromobenzaldehyde

Table 1. Optimization of the Reaction Conditions^a)

^a) Reaction conditions: **1a** (70 mg, 0.52 mmol), **2a** (93 mg, 0.5 mmol), solvent (8 ml) . ^b) Yield of isolated 3a. ^c) The reaction required a prolonged time (17 h) and higher temperature (150°) to complete. ^d) The amount of product decreased with a prolonged reaction time.

was then studied. A Me or MeO group had similar effects (*Table 2*, *Entries 2* and 3). The substrate with an EDG such as MeO at $C(5)$ (*Table 2, Entry 4*) gave an excellent yield, and the same results were obtained with an EWG at $C(5)$ (Table 2, Entries 5 and 6). However, 2h with the potent EWG CF₃ group at $C(5)$ gave a slightly lower yield (Table 2, Entry 7), as did the electron-deficient 2-bromonicotinaldehyde (2l; Table 2, Entry 11). The polysubstituted 2-bromo-4,5-dimethoxybenzaldehyde (2i) was also a suitable substrate, and the reaction could be scaled up to grams without lowering the yields (*Table 2, Entry 8*). The bulky 1-bromonaphthalene-2-carbaldehyde (2k) reacted just as well under the standard conditions as the other substrates (*Table 2, Entry 10*). To our delight, the reaction of the electron-rich heterocyclic 3-bromothiophene-2 carbaldehyde $(2m)$ also went smoothly and gave a high yield of $3m$ (Table 2, Entry 12). Moreover, 2-fluoro- and 2-chlorobenzaldehydes (2n and 2o, resp.) were also tested as the substrates to react with 1a, and the yields were 76 (Table 2, Entry 13) and 61% (Table 2, Entry 14), respectively.

After having investigated the scope for *ortho*-haloarene-carbaldehydes, we then studied the reactions of 2-(2-hydroxyphenyl)acetonitriles. Both substrates with EDGs and EWGs, such as 2-(2-hydroxy-3-methoxyphenyl)acetonitrile (1b), 2-[2-hydroxy-4- (trifluoromethyl)phenyl]acetonitrile (1c), and 2-(5-bromo-2-hydroxyphenyl)acetonitrile $(1d)$, were tested under the optimal reaction conditions $(Table 3)$. Most reactions proceeded with high yields. When 1b was used as the substrate, the structurally

	CN + OH	CHO R. Χ	Cs_2CO_3 (3 equiv.) Cul (0.01 equiv.) 100°, DMF	CN R $3a - 3m$	
	1a	$2b - 2m X = Br$ $2nX = F$ $2oX = Cl$			
Entry	Substrate	Carbaldehyde		Product	Yield $[%]$ ^b)
		R	X		
1	2 _b	$3-F$	Br	3 _b	98
$\overline{2}$	2c	4-Me	Br	3с	86
3	2d	4-MeO	Br	3d	93
$\overline{4}$	2e	5-MeO	Br	3e	98
5	2f	5-F	Br	3f	92
6	2g	5-Cl	Br	3g	95
7	2 _h	$5-CF3$	Br	3h	79
8°)	2i	$4,5-(MeO)_{2}$	Br	3i	98
9	2j	4,5-Dioxole	Br	3j	98
10	2k	1-Bromonaphthalene-2-carbaldehyde		3k	94
11	21	2-Bromopyridine-3-carbaldehyde		31	65
12	2m	3-Bromothiophene-2-carbaldehyde		3m	81
13	2n	H	F	3a	76
14	2 ₀	Η	Cl	3a	61

Table 2. One-Pot Cascade Reaction of 2-(2-Hydroxyphenyl)acetonitrile (1a) with Various ortho-Haloarene-carbaldehydes^a)

a) Reaction conditions: $1a(70 \text{ mg}, 0.53 \text{ mmol})$, $2b - 2m(0.50 \text{ mmol})$, $Cs₂CO₃(488 \text{ mg}, 1.5 \text{ mmol})$, CuI (1 mg, 0.005 mmol), DMF (8 ml), reaction time (most of the reaction were complete within 1 h). ^b) Yield of isolated 3. ^c) The reaction was scaled up to grams (1a (1.33 g, 10 mmol), 2i (2.45 g, 10 mmol), Cs_2CO_3 (9.78 g, 30 mmol), CuI (20 mg, 0.1 mmol), and DMF (20 ml)); 97%.

hindered MeO group of 1b led to a decrease in the yields of the product compared with the corresponding product of 1a (compounds $4a - 4e$, $4h$, $4i$, and $4k$ vs. compounds $3a -$ 3e, 3h, 3i, and 3k), except for 4f, 4g, 4l, and 4m. Yields of 4f, 4g, and 4m were slightly higher than those of $3f$, $3g$, and $3m$, but the yield of $4l$ (92%) was dramatically higher than that of $3I(65\%)$. When 1c was treated with various 2-bromobenzaldehydes, the EWG CF₃ of **1c** also led to a decrease in the yields (compounds 4n, 4o, and 4p vs. compounds 3a, 3g, and 3e). In the case of 1d, substituted 2-bromobenzaldehydes, 2g and 2e, gave higher yields than 2-bromobenzaldehyde (Table 3, Entries 17 and 18 vs. Entry 16). Furthermore, 2-bromobenzaldehyde with a 5-MeO group (Table 3, En $try 15)$ underwent the reaction in excellent yield (95%), while 2-bromobenzaldehyde with a 5-Cl group led to a decreased yield (*Table 3, Entries 14 vs. 13*). As $2-(2$ aminophenyl)acetonitrile $(\mathbf{1d})$ is an analog of $\mathbf{1a}$, $\mathbf{1d}$ was treated with 2-bromobenzaldehyde under the standard conditions without any optimization, to give the corresponding product in only 21% yield (Table 3, Entry 19).

We also tried another strategy to construct dibenzo $[b, f]$ oxepin under the same conditions (0.01 equiv. of CuI, 3 equiv. of Cs₂CO₃, DMF, 100° , 1 h; Scheme 2).

Table 3. One-Pot Cascade Reaction of 2-(ortho-Hydroxy/ortho-Aminophenyl)acetonitriles with Substituted 2-Bromobenzaldehydes^a)

R^1	x 1a $R^1 = H$, $X = OH$ 1b R^1 = 3-MeO, $X = OH$	CN $R^2 \frac{f}{f}$	$Cs2CO3$ (3 equiv.) CHO Cul (0.01 equiv.) 100°, DMF Br $2a - 2i$ Br			CN $R^{1.6}$ $4a - 4i$, $4k - 4s$ $x = 0$ $4t x = N$	
	1c $R^1 = 4 - CF_3$, $X = OH$ 1d R^1 = 5-Br, $X = OH$ 1e $R^1 = H$, $X = NH_2$		CHO 2k	21	CHO CHO Br Br 2m		
Entry	$\mathbf{1}$	\mathbb{R}^1	X	$\boldsymbol{2}$	\mathbb{R}^2	Product	Yield $[%]^{b}$)
\mathcal{I}	$\mathbf b$	3-MeO	OH	a	H	4a	87
\overline{c}	b	$3-MeO$	OH	b	$3-F$	4 _b	83
3	b	$3-MeO$	OH	$\mathbf c$	4-Me	4c	80
$\boldsymbol{4}$	$\mathbf b$	$3-MeO$	OH	d	4-MeO	4d	71
5	b	$3-MeO$	OH	е	5-MeO	4e	90
6	b	3-MeO	OH	$\mathbf f$	$5-F$	4f	98
$\overline{7}$	b	$3-MeO$	OH	g	$5-Cl$	4 _g	97
8	$\mathbf b$	$3-MeO$	OH	h	$5-CF3$	4h	70
9	b	3-MeO	OH	i	$4,5-(MeO)$,	4i	83
10	b	3-MeO	OH	k		4k	80
11	$\mathbf b$	$3-MeO$	OH	L		41	92
12	$\mathbf b$	$3-MeO$	OH	m		4 _m	95
13	c	4 -CF ₃	OH	a	H	4n	72
14	c	4 -CF ₃	OH	g	$5-Cl$	40	63
15	c	4 -CF ₃	OH	e	5-MeO	4p	95
16	d	$5-Br$	OH	a	H	4q	79
17	d	$5-Br$	OH	g	$5-Cl$	4r	92
18	d	$5-Br$	OH	e	5-MeO	4s	86
19	e	H	NH ₂	a	H	4t $[30]$	21

a) Reaction conditions: $1a-1d$ (0.53 mmol), $2a-2m$ (0.50 mmol), Cs_2CO_3 (488 mg, 1.5 mmol), CuI $(1 \text{ mg}, 0.005 \text{ mmol})$, and DMF (8 ml) . b) Yield of isolated 4.

Unfortunately, we did not obtain any desired product. This can be explained by the mechanism of the reaction. We presumed that the cascade includes two steps: the first step is a Knoevenagel condensation, followed by the Ullmann ether formation (Scheme 3). This was established by formation of the same intermediates, as revealed by LC/MS, under Cu-assisted and Cu-free conditions. The intermediate in the formation of 3d was confirmed by ¹H- and ¹³C-NMR, and MS data (Scheme 3). This intermediate could be converted to compound 3d via cyclization in two pathways. When CuI was present, *Ullmann* ether formation occurred subsequently (*Path A*), or the aromatic nucleophilic substitution reaction $(Path B)$ would occur to give 3d under Cu-free condition. However, Path A gave 3d with a preponderant high yield $(92\%; 56\%$

Scheme 2. Another Strategy to Form Dibenzo[b,f]oxepin

for Path B; Table 1, Entries 9 and 14). When salicylaldehyde was used as the substrate in the presence of the base, the phenoxy anion decreased the activity of aldehyde, and Knoevenagel condensation could not occur under our conditions.

Conclusions. – We have developed a simple and efficient method for the synthesis of the pharmacologically important dibenzo $[b, f]$ oxepin scaffold via Cu-assisted/Cufree one-pot tandem reaction. Various substituted 2-(2-hydroxyphenyl)acetonitriles and substituted 2-haloarene-carbaldehydes are tolerated well in the reaction. By this procedure, a library of functionalized dibenzo $[b, f]$ oxepins was achieved quickly with good-to-excellent yields. This approach also provides a practical method, because it could be easily scaled up to grams with excellent yields. Further studies for the synthesis of novel dibenzo $[b, f]$ oxepin analogs and their biological evaluation are currently in progress.

Experiment Part

General. All the reagents, except 1a, are commercially available, and they were used without further purification. Anal. TLC: HSGF 254 (0.15 - 0.2 mm thickness, Yantai Huiyou Company, China). Column chromatography (CC): silica gel (200–300 mesh). M.p. *Büchi 510* melting-point apparatus; uncorrected. ¹H- and ¹³C-NMR spectra: Varian Mercury-300 and/or Varian Mercury-400 spectrometers; δ in ppm rel. to Me₄Si as internal standard, J in Hz. LR- and HR-MS: Finnigan/MAT-95 spectrometer.

2-(2-Hydroxyphenyl)acetonitrile (1a) [29]. The corresponding 2-(2-methoxyphenyl)acetonitrile was dissolved in anh. CH₂Cl₂, followed by slowly adding of $4N$ BBr₃ in CH₂Cl₂ (4 – 5 equiv.) at 0°. The mixture was stirred at r.t. for a few h, then poured onto crushed ice, and the white precipitate was collected by filtration. The product was obtained as white solid.

General Procedure for $3a-3m$, $4a-4i$, and $4k-4s$. To a mixture of 1 (0.53 mmol) and 2bromobenzaldehyde 2 (0.5 mmol) in a 25-ml two-necked reaction flask were added Cs_2CO_3 (1.5 mmol), CuI (0.005 mmol), and DMF (8 ml). The mixture was degassed with Ar (3 \times) and then heated at 100 $^{\circ}$ for 0.5 – 1 h. Most of the reactions were complete within 1 h. Once the reaction was completed (monitored by TLC), the mixture was poured onto crushed ice and then extracted with AcOEt $(3 \times)$. The combined org. layers were washed with brine and dried (Na_2SO_4) . Concentration and purification by CC afforded the products with desirable purities.

3-(2-Bromo-4-methoxyphenyl)-2-(2-hydroxyphenyl)prop-2-enenitrile (the intermediate in *Scheme* 3). Yellow lamellar crystals. M.p. $154-155^{\circ}$. ¹H-NMR (300 MHz, (D_6) DMSO): 10.36 (s, 1 H); 7.96 (d, $J = 8.8$, 1 H); 7.89 (s, 1 H); 7.40 (dd, $J = 7.7$, 1.5, 1 H); 7.35 (d, $J = 2.6$, 1 H); 7.25 (td, $J = 7.8$, 1.6, 1 H); 7.13 (dd, J = 8.8, 2.6, 1 H); 6.97 – 6.87 (m, 2 H); 3.83 (s, 3 H). ¹³C-NMR (125 MHz, (D₆)DMSO): 161.44, 155.55, 144.14 (CH); 130.99 (CH); 130.78 (CH); 129.97 (CH); 126.80, 125.60, 121.26, 120.09 (CH) ; 118.60 (CH); 117.94, 116.82 (CH); 114.54 (CH); 110.70, 56.36 (Me). EI-MS: 329, 331 (40, M⁺), 250 (100, $[M - Br]^+$). HR-EI-MS: 329.0033 (M^+ , C₁₆H₁₂BrNO $_2^+$; calc. 329.0051).

Dibenzo[b,f]oxepine-10-carbonitrile (3a). Compound 3a was obtained after the purification by flash chromatography (FC; SiO₂, 200-300 mesh; petroleum ether (PE)/AcOEt 100:1). Yellowish needles. M.p. 159–160° ([31]: 159–160°). IR (KBr): 2219.67 (CN). ¹H-NMR (300 MHz, (D₆)DMSO): 7.92 (s, 1 H); 7.60 – 7.48 (m, 4 H); 7.42 – 7.27 (m, 4 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 158.00; 157.22; 143.85; 133.90; 132.85; 131.45; 128.18; 126.55; 126.21; 126.12; 122.54; 121.94; 118.76; 112.75. EI-MS: 219 (M⁺, 100). HR-EI-MS: 219.0681 (M^+ , C₁₅H₉NO⁺; calc. 219.0684).

4-Fluorodibenzo[b,f]oxepine-10-carbonitrile $(3b)$. Compound 3b was obtained after the purification by FC (SiO₂; PE/AcOEt 100:1). White powder. M.p. 207–208°. IR (KBr): 2219.67 (CN). ¹H-NMR $(300 \text{ MHz}, (D_6)$ DMSO): 8.00 (s, 1 H); 7.63 – 7.49 (m, 3 H); 7.44 – 7.26 (m, 4 H). ¹³C-NMR (125 MHz, (D_6) DMSO): 156.98; 153.91 $(d, J=248, C-F)$; 144.69 $(d, J=14, C-C-F)$; 142.95 $(d, J=4, C-C-C-F)$; 133.18; 130.51; 128.63; 127.06; 126.84 $(d, J=8, C-C-C-F)$; 126.73 $(d, J=4, C-C-C-F)$; 126.01; 122.46 ; 120.22 $(d, J = 17.6, C-C-F)$; 118.51; 113.70. EI-MS: 237 $(M⁺, 100)$. HR-EI-MS: 237.0593 $(M⁺,$ $C_{15}H_8$ FNO⁺; calc. 237.0590).

3-Methyldibenzo[b,f]oxepine-10-carbonitrile $(3c)$. Compound 3c was obtained after the purification by FC (SiO₂; PE/AcOEt 70:1). Yellowish powder. M.p. 139–140°. IR (KBr): 2213.88 (CN). ¹H-NMR $(300 \text{ MHz}, (\text{D}_6) \text{ DMSO})$: 7.87 (s, 1 H); 7.58 – 7.47 (m, 2 H); 7.36 (m, 3 H); 7.20 (s, 1 H); 7.12 (d, J = 7.7, 1 H); 2.34 (s, 3 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 157.50; 156.60; 144.35; 143.34; 132.17; 130.77; 127.62; 126.38; 125.99; 125.81; 125.00; 122.07; 121.87; 118.44; 111.30; 20.83. EI-MS: 233 (M⁺, 100). HR-EI-MS: 233.0833 (M^+ , C₁₆H₁₁NO⁺; calc. 233.0841).

3-Methoxydibenzo[b,f]oxepine-10-carbonitrile $(3d)$. Compound 3d was obtained after the purification by FC (SiO₂; PE/AcOEt 40:1). Yellowish powder. M.p. $149-150^{\circ}$. IR (KBr): 2211.95 (CN). $1\,\text{H-NMR}$ (300 MHz, (D₆)DMSO): 7.81 (s, 1 H); 7.55 – 7.30 (m, 5 H); 6.99 (d, J = 2.2, 1 H); 6.88 (dd, J = 8.6, 2.4, 1 H); 3.82 (s, 3 H). 13C-NMR (125 MHz, (D6)DMSO): 164.39; 159.47; 156.54; 143.70; 132.66; 132.43; 127.92; 126.56; 126.50; 122.69; 120.92; 119.18; 112.41; 109.88; 107.52; 56.33. EI-MS: 249 (M⁺, 100). HR-EI-MS: 249.0788 (M^+ , C₁₆H₁₁NO $_2^+$; calc. 249.0790).

2-Methoxydibenzo[b,f]oxepine-10-carbonitrile (3e). Compound 3e was obtained after the purification by FC (SiO₂; PE/AcOEt 60:1). Yellowish powder. M.p. $171-172^\circ$. IR (KBr): 2210.02 (CN). $1\,\text{H-NMR}$ (300 MHz, (D_6) DMSO): 7.86 (s, 1 H); 7.58 – 7.48 (m, 2 H); 7.39 – 7.31 (m, 2 H); 7.28 (d, J = 8.4, 1 H); 7.13 – 7.06 (m, 2 H); 3.74 (s, 3 H). 13C-NMR (125 MHz, (D6)DMSO): 157.13; 156.39; 151.27; 143.26; 132.42; 128.29; 127.78; 125.95; 125.63; 122.26; 121.92; 118.72; 118.35; 114.86; 112.66; 55.65. EI-MS: 249 $(M^+, 100)$. HR-EI-MS: 249.0792 $(M^+, C_{16}H_{11}NO_2^+;$ calc. 249.0790).

2-Fluorodibenzo[b,f]oxepine-10-carbonitrile $(3f)$. Compound $3f$ was obtained after the purification by FC (SiO₂; PE/AcOEt 100:1). Yellowish powder. M.p. 218–219°. IR (KBr): 2217.74 (CN). ¹H-NMR $(300 \text{ MHz}, (D_6)$ DMSO): 7.86 $(s, 1 \text{ H})$; 7.63 – 7.47 $(m, 2 \text{ H})$; 7.47 – 7.29 $(m, 5 \text{ H})$. ¹³C-NMR (100 MHz, (D_6) DMSO): 158.86 $(d, J=243, CF)$; 156.77; 153.74; 142.03; 132.68; 129.05 $(d, J=9, C-CH-C-F)$;

 127.88 ; 126.21 ; 125.42 ; 123.18 (d, $J = 9$, CH-CH-C-F); 121.99 ; 119.72 (d, $J = 24$, CH-C-F); 118.01 ; 116 . 68 (d, J = 24, CH–C–F); 113.54. EI-MS 237 (M⁺, 100). HR-EI-MS: 237.0591 (M⁺, C₁₅H₈FNO⁺; calc. 237.0590).

2-Chlorodibenzo[b,f]oxepine-10-carbonitrile $(3g)$. Compound $3g$ was obtained after the purification by FC (SiO₂; PE/AcOEt 100:1). White powder. M.p. $197-198^{\circ}$. IR (KBr): 2217.74 (CN). ¹H-NMR $(300 \text{ MHz}, (\text{D}_6) \text{DMSO})$: 7.86 $(s, 1 \text{ H})$; 7.63 – 7.50 $(m, 4 \text{ H})$; 7.42 – 7.33 $(m, 3 \text{ H})$. ¹³C-NMR (100 MHz, (D6)DMSO): 156.59; 156.23; 141.86; 132.76; 132.72; 130.11; 129.65; 129.32; 127.92; 126.32; 125.37; 123.31; 122.08; 117.98; 113.60. EI-MS: 253 $(M^+$, 100). HR-EI-MS: 253.0302 $(M^+$, C₁₅H₈ClNO⁺; calc. 253.0294).

2-(Trifluoromethyl)dibenzo[b,f]oxepine-10-carbonitrile (3h). Compound 3h was obtained after the purification by FC (SiO₂; PE/AcOEt 100:1). White powder. M.p. 107 – 108°. IR (KBr): 2223.52 (CN). $1H-NMR$ (300 MHz, (D₆)DMSO): 7.99 – 7.89 $(m, 3 H)$; 7.63 – 7.52 $(m, 3 H)$; 7.47 – 7.35 $(m, 2 H)$. ¹³C-NMR (125 MHz, (D_6) DMSO): 160.58; 156.77; 142.43 (CH); 133.36 (CH); 130.65 (d, J = 2.5, $CH-C-CF_3$; 128.95; 128.75 (d, J = 2.5, CH-C-CF₃); 128.48 (CH); 127.05 (CH); 126.68 (d, J = 33, C –CF₃); 125.80; 125.14; 123.25 (CH); 122.98; 122.71 (CH); 118.39. EI-MS: 287 (M⁺, 100). HR-EI-MS: 287.0554 (M^+ , C₁₆H₈F₃NO⁺; calc. 287.0558).

2,3-Dimethoxydibenzo[b,f]oxepine-10-carbonitrile $(3i)$. Compound 3i was obtained after the purification by FC (SiO₂; PE/AcOEt 10:1). Yellow powder. M.p. 157-158°. IR (KBr): 2208.09 (CN). ${}^{1}H\text{-NMR}$ (300 MHz, (D₆)DMSO): 7.76 (s, 1 H); 7.57 – 7.45 (m, 2 H); 7.40 – 7.30 (m, 2 H); 7.07 (s, 1 H); 7.05 (s, 1 H); 3.84 (s, 3 H); 3.74 (s, 3 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 156.87; 153.46; 152.38; 146.45; 143.71; 132.37; 127.77; 126.44; 126.30; 122.43; 119.60; 119.12; 112.87; 110.28; 105.86; 56.41; 56.29. EI-MS: 279 $(M^+, 100)$. HR-EI-MS: 279.0905 $(M^+, C_{17}H_{13}NO_3^+;$ calc. 279.0895).

 $[1,3]$ Dioxolo $[4,5:4,5]$ benzo $[1,2$ -b]benzo $[1]$ oxepine-10-carbonitrile $(=1,3,5$ -Trioxabenzo $[4,5]$ cyclohept[1,2-f]indene-10-carbonitrile; 3j). Compound 3j was obtained after the purification by FC (SiO₂; PE/AcOEt 50:1). Yellow powder. M.p. 199–200°. IR (KBr): 2219.67 (CN). ¹H-NMR (300 MHz, (D_6) DMSO): 7.74 (s, 1 H); 7.57 – 7.45 $(m, 2 H)$; 7.33 $(dd, J = 9.0, 3.0, 2 H)$; 7.08 (s, 1 H); 7.02 (s, 1 H); 6.11 (s, 2 H). 13C-NMR (100 MHz, (D6)DMSO): 156.14; 153.03; 151.60; 144.96; 143.13; 132.18; 127.28; 126.10; 126.06; 121.90; 120.69; 118.61; 110.25; 108.32; 103.16; 102.71. EI-MS: 263 (M^+ , 100). HR-EI-MS: 263.0584 (M^+ , C₁₆H₉NO₃^{*}; calc. 263.0582).

Benzo[b]naphtho[2,1-f]oxepine-8-carbonitrile (3k). Compound 3k was obtained after the purification by FC (SiO2; PE/AcOEt 50 : 1). Yellow powder. M.p. 223–224°. IR (KBr): 2217.74 (CN). ¹H-NMR $(300 \text{ MHz}, (D_6)$ DMSO): 8.67 $(d, J = 7.6, 1 \text{ H})$; 8.07 $(s, 1 \text{ H})$; 8.01 $(d, J = 7.5, 1 \text{ H})$; 7.83 $(d, J = 8.5, 1 \text{ H})$; 7.77 – 7.52 $(m, 6 H)$; 7.42 – 7.348 $(t, J = 7.5, 1 H)$. ¹³C-NMR (100 MHz, (D_6) DMSO): 156.57; 153.20; 143.76; 135.71; 132.73; 128.39; 127.90; 127.80; 127.60; 126.50; 126.22; 125.12; 123.26; 122.71; 122.17; 118.31; 112.84. EI-MS: 269 (M^+ , 100). HR-EI-MS: 269.0839 (M^+ , C₁₉H₁₁NO⁺; calc. 269.0841).

 $Benzo[6,7]oxepino[2,3-b]pyridine-6-carbonitrile (= [1]Benzoxepino[2,3-b]pyridine-6-carbonitrile;$ 3l). Compound 3l was obtained after the purification by FC ($SiO₂$; $PE/ACOE$ 5:1). White powder. M.p. 207 – 208°. IR (KBr): 2219.67 (CN). ¹H-NMR (300 MHz, (D₆)DMSO): 8.43 (d, J = 2.2, 1 H); 8.03 $(d, J = 7.5, 1 \text{ H}); 7.92 (s, 1 \text{ H}); 7.64 - 7.53 (m, 2 \text{ H}); 7.45 - 7.35 (m, 3 \text{ H}).$ ¹³C-NMR (100 MHz, (D₆)DMSO): 161.14; 155.56; 151.70; 141.86; 141.59; 133.34; 128.40; 126.98; 125.48; 123.12; 123.08; 122.77; 118.41; 113.77. EI-MS: 220 (M^+ , 100). HR-EI-MS: 220.0633 (M^+ , C₁₄H₈N₂O⁺; calc. 220.0637).

Benzo[b]thieno[2,3-f]oxepine-9-carbonitrile $(=$ Thieno[3,2-b][1]benzoxepine-9-carbonitrile; $3m$). Compound 3m was obtained after the purification by FC ($SiO₂$; PE/AcOEt 5:1). Yellow powder. M.p. 159 – 160°. IR (KBr): 2213.88 (CN). ¹H-NMR (300 MHz, (D₆)DMSO): 8.00 (d, J = 4.6, 1 H); 7.83 (s, 1 H); 7.61 – 7.43 (m, 2 H); 7.40 – 7.33 (m, 1 H); 7.22 (d, $J = 8.1, 1$ H); 7.02 (d, $J = 4.7, 1$ H). ¹³C-NMR $(100 \text{ MHz}, (D_6)$ DMSO): 158.10; 156.03; 136.89; 133.42; 132.56; 128.39; 126.23; 126.04; 121.92; 121.72; 121.65; 118.51; 109.41. EI-MS: 225 (M^{+} , 100). HR-EI-MS: 225.0240 (M^{+} , C₁₃H₇NOS⁺; calc. 225.0248).

6-Methoxydibenzo[b,f]oxepine-10-carbonitrile (4a). Compound 4a was obtained after the purification by FC (SiO₂; PE/AcOEt 60:1). Yellowish powder. M.p. 171-172°. IR (KBr): 2210.02 (CN). $1\,\text{H-NMR}$ (300 MHz, (D₆)DMSO): 7.93 (s, 1 H); 7.58 – 7.48 (m, 2 H); 7.33 – 7.24 (m, 4 H); 7.06 (m, 1 H); 3.90 (s, 3 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 157.54; 151.89; 144.61; 143.42; 133.24; 130.84; 128.13; 126.85; 126.12; 125.65; 121.69; 118.39; 114.96; 112.35; 56.25. EI-MS: 249 (M^+ , 100). HR-EI-MS: 249.0799 $(M^+$, C₁₆H₁₁NO₂⁺; calc. 249.0790).

4-Fluoro-6-methoxydibenzo[b,f]oxepine-10-carbonitrile (4b). Compound 4b was obtained after the purification by FC (SiO₂: PE/AcOEt 100:1). Yellowish powder. M.p. $248-249^{\circ}$. IR (KBr): 2219.67 (CN). ¹H-NMR (300 MHz, CDCl₃): 7.46 (s, 1 H); 7.28 – 7.23 (m, 1 H); 7.19 – 7.11 (m, 3 H); 7.11 – 7.05 (m, 1 H); 7.02 (d, J = 7.7, 1 H); 3.96 (s, 3 H). ¹³C-NMR (125 MHz, CDCl₃): 155.74 (d, J = 252, CF); 153.65; 147.51; 147.27 $(d, J = 13.9, C-C-F)$; 143.18 $(d, J = 3.8, CH-CH-CH-C-F)$; 132.17; 128.78; 127.58 (CH); 127.16 (d, $J = 7.6$, CH-CH-C-F); 126.69 (d, $J = 3.8$, CH-C-C-C-F); 121.13 (d, $J = 20.2$, CH-C-F); 121.15 (CH); 119.80; 116.57; 116.48 (CH); 58.00 (Me). EI-MS: 267 ($M⁺$, 100). HR-EI-MS: 267.0694 ($M⁺$, $C_{16}H_{10}FNO_2^+$; calc. 267.0696).

6-Methoxy-3-methyldibenzo[b,f]oxepine-10-carbonitrile (4c). Compound 4c was obtained after the purification by FC (SiO₂; PE/AcOEt 20:1). Yellowish powder. M.p. $164-165^{\circ}$. IR (KBr): 2215.81 (CN). $1\,\text{H-NMR}$ (300 MHz, (D₆)DMSO): 7.87 (s, 1 H); 7.38 (d, J = 7.7, 1 H); 7.27 – 7.24 (m, 2 H); 7.12 (s, 1 H); 7.10 $(d, J = 7.7, 1 \text{ H})$; 7.04 $(m, 1 \text{ H})$; 3.90 $(s, 3 \text{ H})$. ¹³C-NMR (100 MHz, (D_6) DMSO): 157.51; 151.91; 144.54; 144.19; 143.44 (CH); 130.66 (CH); 126.98; 126.40 (CH); 126.09 (CH); 125.44; 122.10 (CH); 118.57; 118.36 (CH); 114.80 (CH); 111.38; 56.28 (Me); 20.93 (Me). EI-MS: 263 (M⁺, 100). HR-EI-MS: 263.0949 $(M^+, C_{17}H_{13}NO_2^+;$ calc. 263.0946).

 $3,6$ -Dimethoxydibenzo[b,f]oxepine-10-carbonitrile (4d). Compound 4d was obtained after the purification by FC (SiO₂; PE/AcOEt 40:1). Yellowish powder. M.p. $173-174^{\circ}$. IR (KBr): 2211.95 (CN). $1\,\text{H-NMR}$ (400 MHz, (D₆)DMSO): 7.83 (s, 1 H); 7.44 (d, J = 8.7, 1 H); 7.29 – 7.22 (m, 2 H); 7.02 (dd, J = 6.6, 2.6, 1 H); 6.90 (dd, J = 8.6, 2.6, 1 H); 6.82 (d, J = 2.6, 1 H); 3.89 (s, 3 H); 3.82 (s, 3 H). ¹³C-NMR (100 MHz, (D6)DMSO): 163.66; 158.85; 151.97; 143.91; 143.21 (CH); 132.07 (CH); 127.20; 126.16 (CH); 120.91; 118.76; 118.17 (CH); 114.56 (CH); 111.61 (CH); 109.49; 107.34 (CH); 56.24 (Me); 55.81 (Me). EI-MS: 279 $(M^+, 100)$. HR-EI-MS: 279.0888 $(M^+, C_{17}H_{13}NO_3^+;$ calc. 279.0895).

2,6-Dimethoxydibenzo/b,f/oxepine-10-carbonitrile (4e). Compound 4e was obtained after the purification by FC (SiO₂; PE/AcOEt 40:1). Yellowish powder. M.p. $167-168^{\circ}$. IR (KBr): 2217.74 (CN). ¹H-NMR (400 MHz, (D₆)DMSO): 7.88 (s, 1 H); 7.27 – 7.25 (m, 2 H); 7.22 (d, J = 9.1, 1 H); 7.12 – 7.03 $(m, 3H)$; 3.89 $(s, 3H)$; 3.74 $(s, 3H)$. ¹³C-NMR (100 MHz, (D_6) DMSO): 156.29; 151.76; 151.26; 144.95; 143.28 (CH); 128.61; 126.76; 125.96 (CH); 122.35 (CH); 118.62 (CH); 118.43 (CH); 114.98 (CH); 114.65 (CH); 112.66; 56.22 (Me); 55.64 (Me). EI-MS: 279 (M^+ , 100). HR-EI-MS: 279.0904 (M^+ , $C_{17}H_{13}NO_3^+$; calc. 279.0895).

2-Fluoro-6-methoxydibenzo[b,f]oxepine-10-carbonitrile (4f). Compound 4f was obtained after the purification by FC (SiO₂; PE/AcOEt 50:1). White powder. M.p. $211-212^{\circ}$. IR (KBr): 2217.74 (CN). $1H-NMR$ (300 MHz, $(D_6)DMSO$): 7.85 (s, 1 H); 7.40 – 7.23 (m, 5 H); 7.08 – 7.02 (m, 1 H); 3.89 (s, 3 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 158.77 (d, J = 243, C-F); 153.71; 151.80; 144.54; 142.13; 129.40 (d, $J=9.1, C-CH-C-F$; 126.59; 126.31; 123.25 (d, $J=9.1, CH-CH-C-F$); 119.61 (d, $J=24.3, CH-C-F$); $118.50; 118.11; 116.59 (d, J = 25.2, CH-C-F); 115.29; 113.56; 56.27 (Me). E-I-MS: 267 (M⁺, 100). HR-EI-$ MS: 267.0696 (M^+ , C₁₆H₁₀FNO $_2^+$; calc. 267.0696).

2-Chloro-6-methoxydibenzo[b,f]oxepine-10-carbonitrile $(4g)$. Compound $4g$ was obtained after the purification by FC $(SiO₂; PE/AccEt 50:1)$. Yellowish powder. M.p. 235 – 236°. IR (KBr): 2217.74 (CN). 1 H-NMR (300 MHz, (D6)DMSO): 7.89 (s, 1 H); 7.65 – 7.58 (m, 2 H); 7.35 – 7.28 (m, 3 H); 7.10 – 7.04 (m, 1 H); 3.90 (s, 3 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 156.16; 151.81; 144.35; 141.92 (CH); 132.64 (CH); 130.00 (CH); 129.67; 129.55; 126.54; 126.42 (CH); 123.37 (CH); 118.53 (CH); 118.08; 115.32 (CH); 113.61; 56.29 (Me). EI-MS: 283 (M^+ , 100). HR-EI-MS: 283.0402 (M^+ , C₁₆H₁₀ClNO $_2^+$; calc. 283.0400).

6-Methoxy-2-(trifluoromethyl)dibenzo[b,f]oxepine-10-carbonitrile (4h). Compound 4h was obtained after the purification by FC (SiO₂; PE/AcOEt 50:1). White powder. M.p. $180-181^\circ$. IR (KBr): 2219.67 (CN). ¹H-NMR (300 MHz, (D₆)DMSO): 7.97 (s, 1 H); 7.93 (s, 1 H); 7.88 (dd, J = 8.3, 1 H); 7.47 (d, J = 8.4, 1 H); 7.31 – 7.27 (m, 2 H); 7.10 – 7.02 (m, 1 H); 3.90 (s, 3 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 160.01 ; 151.89 ; 144.00 ; 141.96 ; 130.00 $(d, J = 3.0, CH - C - CF_3)$; 128.82 ; 128.14 $(d, J = 4.0, CH - C - CF_3)$; 126.66 ; 126.38 (d, $J = 32$, C-CF₃); 126.51 ; 123.59 (d, $J = 270$, CF₃); 122.84 ; 118.57 ; 117.98 ; 115.47 ; 113.89 ; 56.33 (Me). EI-MS: 317 (M⁺, 100). HR-EI-MS: 317.0663 (M⁺, C₁₇H₁₀F₃NO₂⁺; calc. 317.0664).

2,3,6-Trimethoxydibenzo[b,f]oxepine-10-carbonitrile (4i). Compound 4i was obtained after the purification by FC (SiO₂; PE/AcOEt 10:1). Yellow powder. M.p. 208-209°. IR (KBr): 2211.95 (CN). $1\,\text{H-NMR}$ (300 MHz, (D₆)DMSO): 7.79 (s, 1 H); 7.28 – 7.24 (m, 2 H); 7.11 (s, 1 H); 7.04 – 7.00 (m, 1 H); 6.87 (s, 1 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 152.87; 151.90; 151.84; 146.10; 144.41; 143.36 (CH); 127.19; 125.99 (CH); 119.66; 118.82; 118.17 (CH); 114.66 (CH); 112.35 (CH); 109.99; 105.32 (CH); 56.29 (Me); 55.92 (Me); 55.92 (Me). EI-MS: 309 (M^+ , 100). HR-EI-MS: 309.0998 (M^+ , C₁₈H₁₅NO₄^{*}; calc. 309.1001).

12-Methoxybenzo[b]naphtho[2,1-f]oxepine-8-carbonitrile (4k). Compound 4k was obtained after the purification by FC (SiO₂; PE/AcOEt 50:1). Yellow power. M.p. $172-173^{\circ}$. IR (KBr): 2217.74 (CN). $1\,\text{H-NMR}$ (300 MHz, (D_6) DMSO): 9.03 $(d, J = 8.4, 1 \text{ H})$; 8.00 $(s, 1 \text{ H})$; 7.95 $(d, J = 8.4, 1 \text{ H})$; 7.80 $(d, J = 1)$ $8.5, 1 \text{ H}$; $7.73 - 7.63$ $(m, 2 \text{ H})$; 7.51 $(d, J = 8.5, 1 \text{ H})$; $7.32 - 7.28$ $(m, 2 \text{ H})$; $7.12 - 7.05$ $(m, 1 \text{ H})$; 3.96 $(s, 3 \text{ H})$. ¹³C-NMR (100 MHz, (D₆)DMSO): 154.44; 151.91; 145.61; 144.25 (CH); 135.74; 128.35 (CH); 127.46 (CH); 127.35; 126.98 (CH); 126.83; 126.66 (CH); 126.53 (CH); 125.26 (CH); 123.99 (CH); 123.54; 118.72 (CH); 118.47; 115.17 (CH); 113.00; 56.16 (Me). EI-MS: 299 (M^+ , 100). HR-EI-MS: 299.0948 (M^+ , $C_{20}H_{13}NO_2^+$; calc. 299.0946).

 $10-Methoxybenzo[6,7]oxepino[2,3-b]pyridine-6-carbonitrile (=10-Methoxy[1]benzoxepino[2,3-d]ba-8-carbonitrile (=10-Methoxy[1]benzoxepino[2,3-d]ba-8-carbonitrile = 10-Methoxy[1]benzoxepino[2,3-d]ba-8-carbonitrile = 10-Methoxy[1]benzoxepino[2,3-d]ba-8-carbonitrile = 10-Methoxy[1]benzoxepino[2,3-d]ba-8-carbonitrile = 10-Methoxy[1]benzoxepino[2,3-d]ba-8-carbonitrile = 10-Methoxy[1]benzoxepino[2,$ b]*pyridine-6-carbonitrile*; **4l**). Compound **4l** was obtained after the purification by FC (SiO₂; PE/AcOEt 5:1). Yellowish powder. M.p. 152–153°. IR (KBr): 2221.59 (CN). ¹H-NMR (300 MHz, (D₆)DMSO): $8.44 (d, J = 4.0, 1 \text{ H})$; $8.04 (d, J = 7.4, 1 \text{ H})$; $7.92 (s, 1 \text{ H})$; $7.45 (t, J = 6.0, 1 \text{ H})$; $7.33 (d, J = 4.2, 2 \text{ H})$; $7.10 (m,$ 1 H); 3.90 (s, 3 H). ¹³C-NMR (125 MHz, (D₆)DMSO): 161.61; 152.70; 151.52 (CH); 144.10; 142.06 (CH); 141.50 (CH); 127.13 (CH); 126.66; 123.10; 122.97 (CH); 119.16 (CH); 118.55; 116.21 (CH); 113.90; 56.81 (Me). EI-MS: 250 (M^+ , 100) HR-EI-MS: 250.0736 (M^+ , $C_{15}H_{10}N_2O_2^+$; calc. 250.0742).

 $5-Methoxybenzo/b]$ thieno[2,3-f]oxepine-9-carbonitrile $(=5-Methoxythieno[3,2-b]/1]$ benzoxepine-9-carbonitrile; $4m$). Compound $4m$ was obtained after the purification by FC (SiO₂; PE/AcOEt 5:1). Yellow powder. M.p. $180 - 181^\circ$. IR (KBr): 2211.95 (CN). ¹H-NMR (300 MHz, (D₆)DMSO): 7.95 (*d, J* = $5.4, 1 \text{ H}$; 7.86 (d, $J = 0.5, 1 \text{ H}$); $7.34 - 7.25$ (m, 2 H); $7.07 - 6.99$ (m, 1 H); 6.96 (dd, $J = 5.4, 0.5, 1 \text{ H}$); 3.87 (s, 3 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 158.25; 151.64; 144.11; 137.01 (CH); 133.41 (CH); 127.38; 126.30 (CH); 122.48; 121.55 (CH); 119.06 (CH); 118.65; 115.42 (CH); 109.54; 56.28 (Me). EI-MS: 255 $(M^+, 100)$. HR-EI-MS: 255.0354 $(M^+, C_{14}H_9NO_2S^+;$ calc. 255.0354).

7-(Trifluoromethyl)dibenzo[b,f]oxepine-10-carbonitrile (4n). Compound 4n was obtained after the purification by FC (SiO₂; PE/AcOEt 100:1). Yellow powder. M.p. 108-110°. IR (KBr): 2219.67 (CN). 1 H-NMR (300 MHz, (D6)DMSO): 8.10 – 8.01 (m, 1 H); 7.88 – 7.68 (m, 3 H); 7.66 – 7.46 (m, 3 H); 7.42 – 7.30 (m, 1 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 157.16; 156.57; 145.50 (CH); 138.35; 133.99 (CH); 132.08 (d, $J = 32$, $C-CF_3$); 131.31 (CH); 129.74; 129.00 (CH); 126.02 (d, $J = 273$, CF₃); 126.17 (CH); $122.87 (d, J = 3, CH-C-CF₃); 121.69 (CH); 119.36 (CH-C-CF₃); 117.86; 114.52. EI-MS: 287 (M⁺, 100).$ HR-EI-MS: 287.0554 (M^+ , C₁₆H₈F₃NO⁺; calc. 287.0558).

2-Chloro-7-(trifluoromethyl)dibenzo[b,f]oxepine-10-carbonitrile (4o). Compound 4o was obtained after the purification by FC (SiO₂; PE/AcOEt 100 : 1). Yellow powder. M.p. $176 - 177^\circ$. IR (KBr): 2221.59 (CN). ¹H-NMR (400 MHz, (D₆)DMSO): 8.02 (s, 1 H); 7.89 (s, 1 H); 7.76 – 7.73 (m, 2 H); 7.69 – 7.64 (m, 2 H); 7.52 (d, $J = 8.3$, 1 H). ¹³C-NMR (100 MHz, (D_6) DMSO): 161.59; 161.03; 149.21 (CH); 138.52 (CH); 137.52 (d, J = 32, C-CF₃); 135.59 (CH); 135.28; 134.62; 134.41 (CH); 134.14; 128.43 (d, J = 271, CF_3); 128.77 (CH); 128.32 (d, J = 4.0, CH–C–CF₃); 124.65 (d, J = 4.0, CH–C–CF₃); 122.78; 117.65. EI-MS: 321 $(M^+, 100)$. HR-EI-MS: 321.0159 $(M^+, C_{16}H_7CIF_3NO^+$; calc. 321.0168).

2-Methoxy-7-(trifluoromethyl)dibenzo[b,f]oxepine-10-carbonitrile (4p). Compound 4p was obtained after the purification by FC (SiO₂: PE/AcOEt 50:1). Yellowish powder. M.p. $181-182^\circ$. IR (KBr): 2221.59 (CN). ¹H-NMR (400 MHz, (D₆)DMSO): 8.02 (s, 1 H); 7.83 (s, 1 H); 7.75 – 7.68 (m, 2 H); 7.40 (d, $J = 8.8, 1 \text{ H}$); 7.16 (dd, $J = 8.8, 3.1, 1 \text{ H}$); 7.12 (d, $J = 3.1, 1 \text{ H}$); 3.75 (s, 3 H). ¹³C-NMR (100 MHz, (D_6) DMSO): 156.92; 156.67; 150.83; 145.38 (CH); 132.10 (d, $J=33$, C-CF₃), 129.69; 129.06 (CH); 127.94; 123.32 $(d, J = 272, CF_3)$; 122.74 $(d, J = 3.0, CH - C - CF_3)$; 122.51 (CH) ; 119.25 (CH) ; 119.17 $(d, J = 3.0, CH - C - CF_3)$ $J = 3.0, \, \text{CH--C--CF}_3$); 117.89; 115.15 (CH); 111.51; 55.71 (Me). EI-MS: 317 (M⁺, 100). HR-EI-MS: 317.0656 (M^+ , C₁₇H₁₀F₃NO₂⁺; calc. 317.0664).

8-Bromodibenzo[b,f]oxepine-10-carbonitrile (4q). Compound 4q was obtained after the purification by FC (SiO₂; PE/AcOEt 100:1). Light yellow powder. M.p. $172-173^{\circ}$. ¹H-NMR (300 MHz, $(D₆)$ DMSO): 7.98 (s, 1 H); 7.72 (dd, J = 8.6, 2.4, 1 H); 7.61 – 7.53 (m, 2 H); 7.51 (dd, J = 7.7, 1.3, 1 H); $7.39 - 7.33$ (m, 2 H); 7.31 (td, J = 7.6, 1.2, 1 H). ¹³C-NMR (125 MHz, (D₆)DMSO): 157.79; 156.44; 145.19 (CH); 135.44 (CH); 134.31 (CH); 131.74 (CH); 130.34 (CH); 128.33; 127.96; 126.53 (CH); 124.82 (CH);

122.01 (CH); 118.42; 111.46. EI-MS: 297, 299 (M^+ , 100). HR-EI-MS: 296.9789 (M^+ , C₁₅H₈BrNO⁺; calc. 296.9789).

8-Bromo-2-chlorodibenzo[b,f]oxepine-10-carbonitrile (4r). Compound 4r was obtained after the purification by FC (SiO₂; PE/AcOEt 100:1). White powder. M.p. $194-195^{\circ}$. ¹H-NMR (300 MHz, (D_6) DMSO): 7.92 (s, 1 H); 7.75 (dd, J = 8.1, 1.8, 1 H); 7.65 – 7.57 (m, 3 H); 7.43 – 7.35 (m, 2 H). ¹³C-NMR (125 MHz, (D₆)DMSO): 156.47; 156.27; 143.68 (CH); 135.77 (CH); 133.64 (CH); 130.83 (CH); 130.53 (CH) ; 130.42; 129.54; 128.00; 124.84 (CH); 123.88 (CH); 118.68; 118.12; 112.76. EI-MS: 331, 333 (M^{+} , 100). HR-EI-MS: 330.9399 (M^+ , C₁₅H₇BrClNO⁺; calc. 330.9400).

8-Bromo-2-methoxydibenzo[b,f]oxepine-10-carbonitrile (4s). Compound 4s was obtained after the purification by FC (SiO₂; PE/AcOEt 100 : 1). Light yellow powder. M.p. 175 – 176°. ¹H-NMR (300 MHz, $(D₆)$ DMSO): 7.91 (s, 1 H); 7.71 (dd, J = 8.6, 2.3, 1 H); 7.56 (d, J = 2.3, 1 H); 7.33 (d, J = 8.6, 1 H); 7.28 (d, $J = 8.7, 1$ H); 7.10 (m, 2 H); 3.72 (s, 3 H). ¹³C-NMR (125 MHz, (D₆)DMSO): 157.04; 156.77; 151.45; 145.03 (CH); 135.44 (CH); 130.37 (CH); 128.49; 128.26; 124.64 (CH); 122.78 (CH); 119.53 (CH); 118.41; 118.22; 115.57 (CH); 111.78; 56.17 (Me). EI-MS: 327, 329 (M⁺, 100). HR-EI-MS: 326.9896 (M⁺, $C_{16}H_{11}BrNO_2^+$; calc. 326.9895).

5H-Dibenzo/b,f/azepine-10-carbonitrile (4t) [30]. Compound 4t was obtained after the purification by FC (SiO₂; PE/AcOEt 3:1). Brown powder. M.p. 218–219°. IR (KBr): 2217.74 (CN). ¹H-NMR $(300 \text{ MHz}, (D_6) \text{ DMSO})$: 12.61 (s, 1 H); 8.03 – 7.96 (m, 2 H); 7.67 – 7.51 (m, 5 H); 7.33 (td, J = 7.2, 1.1, 1 H); 7.26 (td, J = 7.2, 1.1, 1 H). ¹³C-NMR (100 MHz, (D₆)DMSO): 144.74; 135.51; 129.98 (CH); 129.33 (CH); 129.33 (CH); 129.33; 128.26; 126.97 (CH); 126.97 (CH); 123.93 (CH); 122.05 (CH); 118.37 (CH); 117.00; 112.66 (CH). EI-MS: 218 (M^+ , 100). HR-EI-MS: 218.0846 (M^+ , $C_{15}H_{10}N_2^+$; calc. 218.0844).

This work was financially supported by the 'Interdisciplinary Cooperation Team' Program for Science and Technology Innovation of the Chinese Academy of Sciences, the National Natural Science Foundation of China (21072205, SIMM1203ZZ-0103, and SIMM1105KF-02).

REFERENCES

- [1] E. Tojo, D. Dominguez, L. Castedo, Phytochemistry 1991, 30, 1005.
- [2] T.-X. Qian, L.-N. Li, Phytochemistry 1992, 31, 1068.
- [3] Y.-H. Lu, C.-N. Lin, H.-H. Ko, S.-Z. Yang, L.-T. Tsao, J.-P. Wang, *Helv. Chim. Acta* 2003, 86, 2566.
- [4] P. Kittakoop, S. Nopichai, N. Thongon, P. Charoenchai, Y. Thebtaranonth, Helv. Chim. Acta 2004, 87, 175.
- [5] L.-H. Mu, J.-B. Li, J.-Z. Yang, D.-M. Zhang, J. Asian Nat. Prod. Res. 2007, 9, 649.
- [6] G. R. Pettit, A. Numata, C. Iwamoto, Y. Usami, T. Yamada, H. Ohishi, G. M. Cragg, J. Nat. Prod. 2006, 69, 323.
- [7] D. Acton, G. Hill, B. S. Tait, J. Med. Chem. 1983, 26, 1131.
- [8] W. S. Riehen, H. B. Basel, US Pat. 3641056, 1972.
- [9] A. A. Trabanco, J. M. Alonso, J. I. Andérs, J. M. Cid, J. Fernández, L. Iturrino, A. Megens, Chem. Pharm. Bull. 2004, 52, 262.
- [10] M. N. Agnew, A. Rizwaniuk, H. H. Ong, J. K. Wichmann, J. Heterocycl. Chem. 1986, 23, 265.
- [11] K. Hino, H. Nakamura, S. Kato, A. Irie, Y. Nagai, H. Uno, Chem. Pharm. Bull. 1988, 36, 3462. [12] R. Rupčić, M. Modrić, A. Hutinec, A. Čikoš, B. Stanić, M. Mesić, D. Pešić, M. Merćep, J. Heterocycl.
- Chem. 2010, 47, 640. [13] I. Ueda, Y. Sato, S. Maeno, S. Umio, Chem. Pharm. Bull. 1978, 26, 3058.
- [14] J. Fernández, J. M. Alonso, J. I. Andrés, J. M. Cid, A. Díaz, L. Iturrino, P. Gil, A. Megens, V. K. Sipido, A. A. Trabanco, J. Med. Chem. 2005, 48, 1709.
- [15] A. A. Trabanco, J. M. Alonso, J. M. Cid, L. M. Font, A. Megens, Farmaco 2005, 60, 241.
- [16] R. Kiyama, T. Honma, K. Hayashi, M. Ogawa, M. Hara, M. Fujimoto, T. Fujishita, J. Med. Chem. 1995, 38, 2728.
- [17] S. Jinno, T. Okita, Heterocycles 1999, 51, 303.
- [18] P. A. C. Cloos, F. Reissig, P. Boissy, M. Stahlhut, PCT Int. Appl. WO2004039773.
- [19] K. Zimmermann, S. Roggo, E. Kragten, P. Fürst, P. Waldmeier, Bioorg. Med. Chem. Lett. 1998, 8, 1195.
- [20] K. Zimmermann, P. C. Waldmeier, W. G. Tatton, Pure Appl. Chem. 1999, 71, 2039.
- [21] P. G. Nantermet, H. A. Rajapakse, PCT Int. Appl. WO2007019080.
- [22] R. H. F. Manske, A. E. Ledingham, J. Am. Chem. Soc. 1950, 72, 4797.
- [23] F. A. L. Anet, P. M. G. Bavin, Can. J. Chem. 1957, 35, 1084.
- [24] R. Olivera, R. SanMartin, F. Churruca, E. Domínguez, J. Org. Chem. 2002, 67, 7215.
- [25] Z. Yang, H. N. C. Wong, P. M. Hon, H. M. Chang, C. M. Lee, J. Org. Chem. 1992, 57, 4033.
- [26] T. Storz, E. Vangrevelinghe, P. Dittmar, Synthesis 2005, 15, 2562.
- [27] Z. Cong, T. Miki, O. Urakawa, H. Nishino, J. Org. Chem. 2009, 74, 3978.
- [28] N. B. Chernysheva, A. V. Samet, V. N. Marshalkin, V. A. Polukeev, V. V. Semenov, Mendeleev Commun. 2001, 11, 109.
- [29] Y. Chen, H. Xiang, Y. Xie, C. Yang, Heterocycles 2011, 83, 1355.
- [30] H. Blattner, A. Storni, Pat. EP 0011603 A1, 1980.
- [31] P. M. G. Bavin, K. D. Bartle, D. W. Jones, J. Heterocycl. Chem. 1968, 5, 327.

Received March 8, 2012